



A 34-year-old woman is brought to the emergency department by ambulance due to nausea, an episode of vomiting, and dizziness. Her boyfriend arrives with her and says that he found her lying on the bed next to an empty bottle of aspirin tablets. When asked about the potential ingestion, the patient admits to swallowing a "bunch of aspirin pills a little more than 12 hours ago" after leaving work yesterday. She also has "an annoying buzzing sound that won't stop." Temperature is 38.3 C (101 F) and pulse is 102/min. She appears agitated and confused. Which of the following sets of laboratory results would most likely be found in this patient?

	рН	PaCO <sub>2</sub>	Plasma HCO <sub>3</sub>
<b>A</b> .	7.30	34 mm Hg	16 mEq/L
B.	7.38	20 mm Hg	12 mEq/L
C.	7.42	40 mm Hg	26 mEq/L
D.	7.48	50 mm Hg	36 mEq/L
) E.	7.60	25 mm Hg	24 mEq/L

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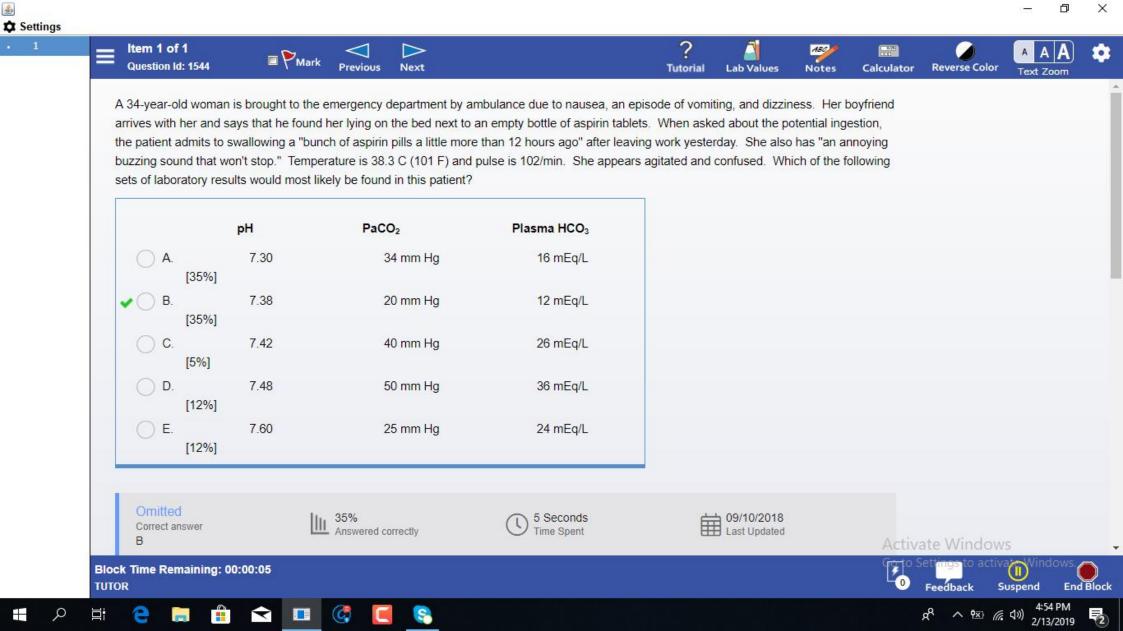
































X

Acute salicylate intoxication typically presents with nausea/vomiting, confusion, dizziness, tinnitus (eg, buzzing sound), fever, and tachypnea within several hours after ingestion. Two different acid-base abnormalities are characteristic.

- 1. Respiratory alkalosis occurs first as salicylates directly stimulate the medullary respiratory center, resulting in increased ventilation and loss of CO2 in the expired air.
- 2. Anion gap metabolic acidosis begins to develop about 12 hours later as high concentrations of salicylates increase lipolysis, uncouple oxidative phosphorylation, and inhibit the citric acid cycle. This results in the accumulation of organic acids in the blood (eg. ketoacids, lactate, pyruvate), which bind bicarbonate, thereby increasing the calculated anion gap.

Because this patient is presenting >12 hours after salicylate ingestion, it is likely that the initial primary respiratory alkalosis has progressed to mixed respiratory alkalosis/metabolic acidosis. The arterial blood gas (ABG) in salicylate toxicity often shows a pH in the normal range as the 2 primary acid-base disturbances shift the pH in opposite directions. In addition, the ABG will show a low PaCO<sub>2</sub> (due to respiratory alkalosis and compensation for metabolic acidosis) and low HCO<sub>3</sub> (due to metabolic acidosis). The PaCO<sub>2</sub> will be lower than predicted respiratory compensation (in this case, expected PaCO<sub>2</sub> = [1.5 \* 12] + 8 ± 2 = 26 ± 2 = 24-28 mm Hg) due to the concurrent primary respiratory alkalosis.

(Choice A) These values represent a primary metabolic acidosis with appropriate respiratory compensation (eg, expected PaCO<sub>2</sub> = [1.5 \* 16] + 8 ± 2 = 32 ± 2 = 30-34 mm Hg). Compensatory responses do not correct the pH completely.

(Choice C) These values are indicative of a normal ABG.

(Choice D) These values represent a primary metabolic alkalosis with respiratory compensation. Although patients with significant vomiting may develop a mild contraction metabolic alkalosis, the metabolic acidosis due to salicylate intoxication would predominate.

(Choice E) These values represent a primary respiratory alkalosis, which can be seen in early salicylate intoxication; they would be unlikely >12 hours after a massive ingestion.

## Educational objective:

Aspirin intoxication should be suspected in a patient with the triad of fever, tinnitus, and tachypnea. Adults with aspirin toxicity initially develop a Activate Windows

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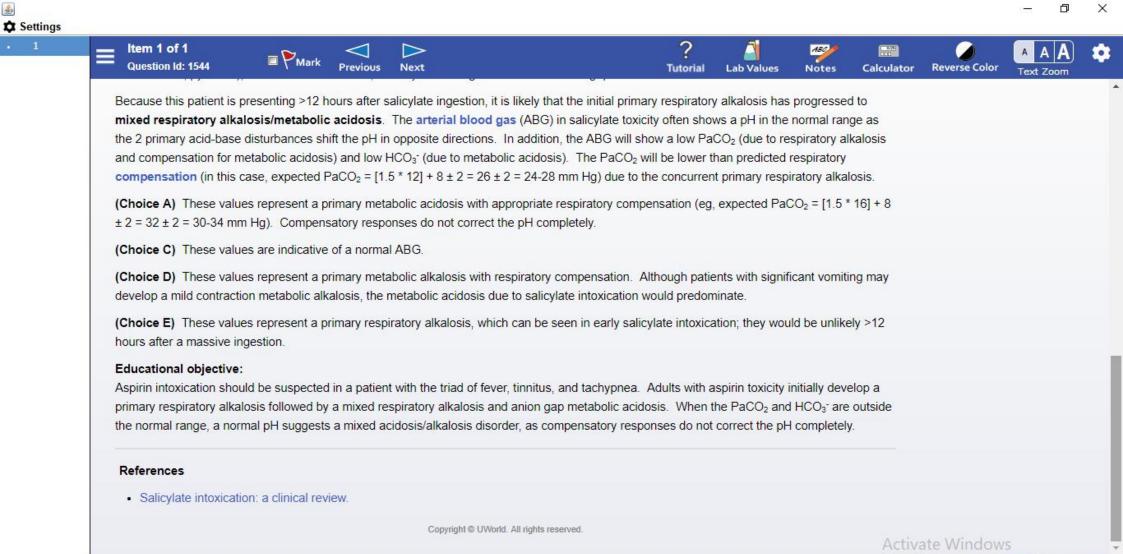












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